

# Reading the Periodic Table

3

Li

Lithium

6.941

3

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Lithium

6.941

← Element Symbol

← Element Name



## Atomic Number

The atomic number tells you the number of PROTONS.

It is also the number of electrons.

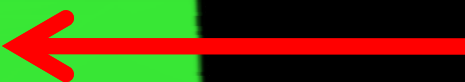
3

Li

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6.941

Atomic Mass



6.941

What does Atomic Mass  
tell us?

Protons  
+ Neutrons

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Atomic Mass  
(Rounded)

3  
Li

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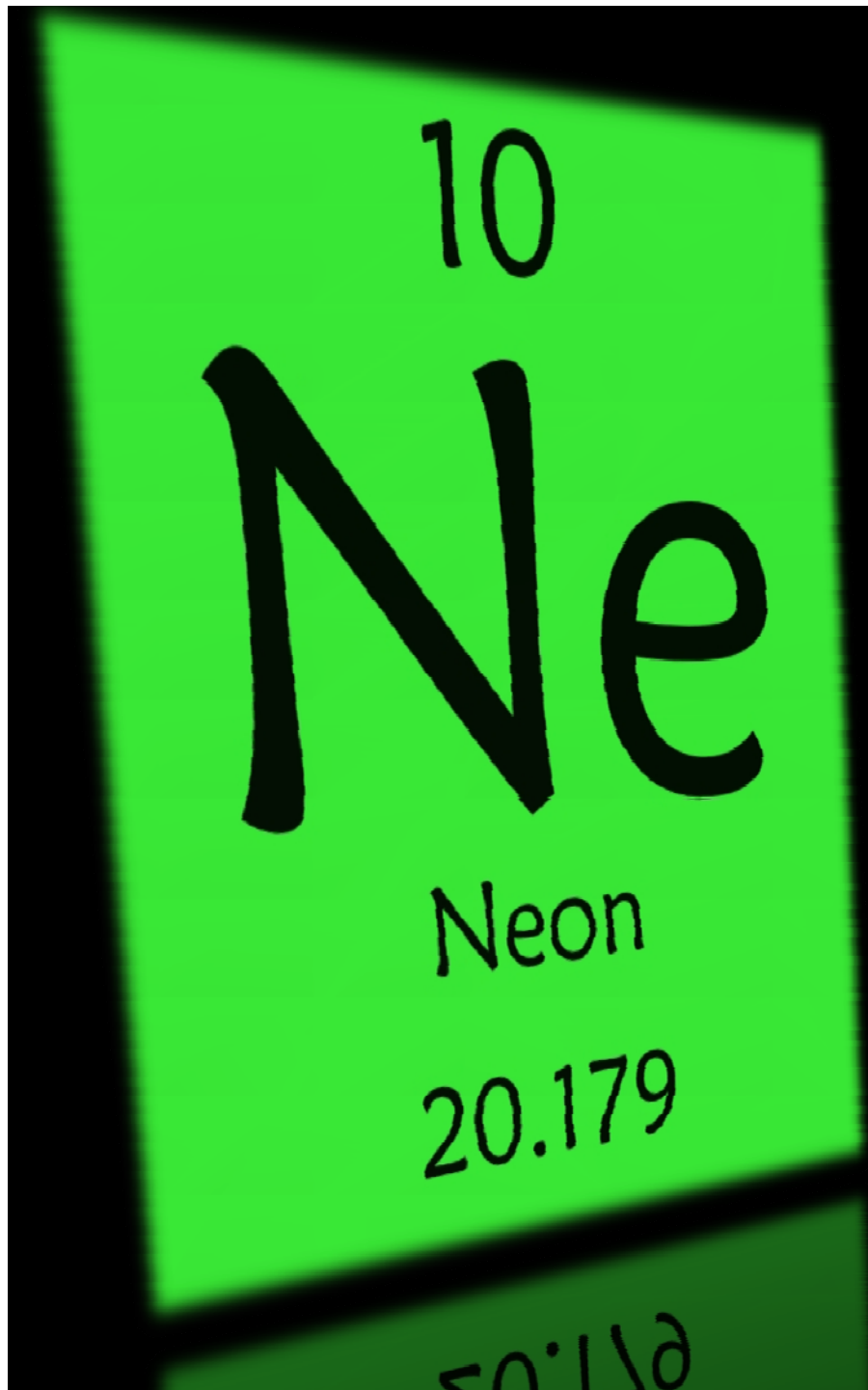
6.941

# QUESTION...

How could you find the number of neutrons???

Answer: Atomic mass (rounded)

$$\frac{\text{Atomic \#}}{\text{\# of neutrons}}$$



Lets give it a Try!

How many protons,  
electrons, and  
neutrons does Ne  
have?

10  $p^+$

10  $e^-$

10  $n^0$



# Let's Practice...

Find the number of  $p^+$ ,  $e^-$ , and  $n$  for each element:

1. H

1  $p^+$

1  $e^-$

0  $n$

2. He

2  $p^+$

2  $e^-$

2  $n$

3. Tb

65  $p^+$

65  $e^-$

94  $n$

# HOMEWORK...

Find the number of  $p^+$ ,  $e^-$ , and  $n$  for each element:

1. Si

2. Hg

3. Fe

4. Ta

5. Pt